

RING-CURRENT EFFECTS ON CARBON-13 CHEMICAL SHIFTS OF
SP-HYBRIDIZED CARBONS IN CYCLOPHADIYNES¹⁾

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Ring-current effects have been important, familiar phenomena in the ¹H NMR spectroscopy. However, there have been rather a few studies of ring-current effects on ¹³C-chemical shifts.²⁾ Model compounds needed for such studies should be selected with adequate prudence because the ¹³C resonance shift caused by a ring-current effect is generally too small as compared with the ¹³C-chemical shift variation due to other factors. Recently, marked ring-current effects on ¹³C-chemical shifts were observed in some bridged annulenes.^{2a,b)} For benzenoid system, weak effects were observed on ¹³C NMR study of [12]paracyclophane. though the methylene-carbon signals in problem were not clearly assigned.^{2c)}

We have studied ¹³C NMR spectra of cyclophadiynes³⁾ from the following advantage: in each pair of rigid compounds A₃ and B₃, and strainless ones A₄ and B₄, the diacetylene sp-carbons should be closely fixed with the same spatial arrangements on the aromatic ring as seen from X-ray analysis of B₃,^{3b)} and were considered to be appropriate ones for examining ring-current effects on ¹³C-chemical shifts. Thus, we have found clear evidence for the presence of these effects on their ¹³C NMR spectra. The ¹³C shifts obtained are shown in the Figure.⁴⁾

Both middle and propargylic methylene-carbons in A₃ show almost equal chemical shifts to those of the corresponding carbons in B₃. On the other hand, both inner and outer sp-carbons in A₃ are shielded by 1.2ppm relative to the corresponding carbons in B₃. This upfield shifts of 1.2ppm should be mainly caused by the difference in ring-current effects between the two aromatic rings, which was theoretically estimated earlier.⁵⁾ Similar but slight upfield shifts of sp-carbons (inner -0.4; outer -0.6ppm) are also observed in the tetramethylene-bridged pair,

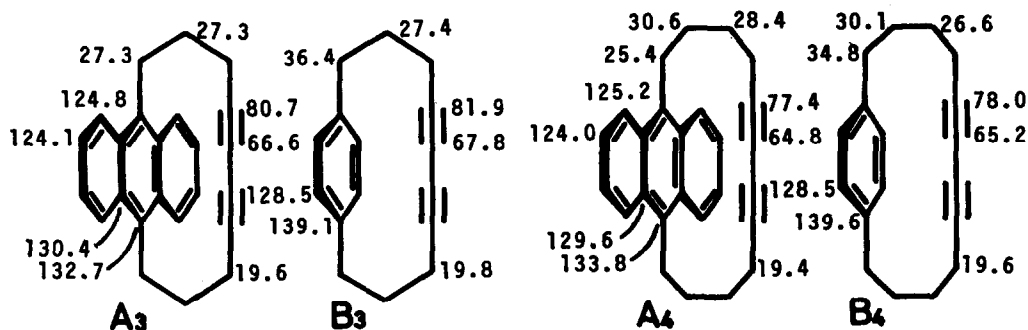


Figure. Carbon-13 chemical shifts, δ

A₄ and B₄. The larger upfield shifts seen in the trimethylene-bridged pair than those observed in the tetramethylene-bridged one are easily explained by the shorter non-bonded interatomic distances between the sp -carbons and the aromatic rings in the former pair.

In comparison of the chemical shifts of strained B₃ with those of strainless B₄, downfield shifts of sp -carbons (inner +2.6; outer +3.9ppm) and an upfield shift of the bridgehead sp^2 -carbons (-0.5ppm) are found. These shifts would be caused by rehybridization of normal sp - or sp^2 -hybridized carbons due to trans-annular π -electron interactions containing repulsion. Similar shifts of sp -carbon signals and the bridgehead sp^2 -ones are also found in the anthracenophadiyne pair, A₃ and A₄.

References and Footnote

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